

Water-soluble copper(II) complexes copper(II) with polyethylenaminomethylphosphonic acid

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Abstract

In present paper the equilibrium of polyethylenepolyamine-*N*-methylphosphonic acid in water solutions was studied, thermodynamic characteristics of the acid-base equilibrium, formation of polyprotonation and normal complex compounds with ions of copper(II) have been obtained. Enthalpy complexing in various concentrations of copper(II) ions and pH of the initial solution was measured by calorimetric method.