

Investigation of the potential energy surface of bimolecular reactions of nucleophilic substitution on the base of density functional theory

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Abstract

Analysis of the thermodynamic parameters of bimolecular reactions of nucleophilic substitution in the gas phase and in solution using density functional calculations in the program packages GAUSSIAN and ADF using a number of all-electron basis sets has been carried out. The geometry of the transition state, and early- and last transition states for the reaction $\text{CH}_3\text{Cl} + \text{F}^- \rightarrow \text{CH}_3\text{F} + \text{Cl}^-$ have been calculated. Comparative analysis of the energy profiles of reaction in the gas phase, water and carbon tetrachloride was carried out.