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## Investigation of the potential energy surface of bimolecular reactions of nucleophilic substitution on the base of density functional theory

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## **Abstract**

Analysis of the thermodynamic parameters of bimolecular reactions of nucleophilic substitution in the gas phase and in solution using density functional calculations in the program packages GAUSSIAN and ADF using a number of all-electron basis sets has been carried out. The geometry of the transition state, and early-and last transition states for the reaction  $CH_3Cl + F \rightarrow CH_3F + Cl$  have been calculated. Comparative analysis of the energy profiles of reaction in the gas phase, water and carbon tetrachloride was carried out.