

Phase equilibriums in systems with involvement of *n*-alkanes, perchloromethane and perluxe

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Abstract

Liquiduses of binary systems $\text{CCl}_4 - n\text{-C}_n\text{H}_{2n+2}$ and $\text{C}_2\text{Cl}_4 - n\text{-C}_n\text{H}_{2n+2}$ ($n = 10-20$) by method UNIFAC and by means of Schroder–Le-Shatele's equation are counted. Phase equilibriums in the pointed systems are experimentally learnt. Comparison rated and observational methods is spent. It is demonstrated, that with magnification n in the *n*-alkane molecule content decrease $n\text{-C}_n\text{H}_{2n+2}$ in eutectic compositions is observed, and melting point of eutectics slightly differ from melting points CCl_4 and C_2Cl_4 .