

Synthesis and structure of the solvates bis(μ_2 -dimethylarsinato)-(μ_2 -oxo)-bis(triphenylantimony) with benzene and dimethylsulfoxide

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Keywords: pentaphenylantimony, dimethylarsine acid, oxygen.

Abstract

By interaction of pentaphenylantimony with dimethylarsine acid in the presence of oxygen in benzene or dimethylsulfoxide (dmsO) we synthesized solvates of bis(μ_2 -dimethylarsinato)-(μ_2 -oxo)-bis(triphenylantimony) with benzene $[(\text{Ph}_3\text{Sb})_2(\mu_2\text{-O}_2\text{AsMe}_2)_2(\mu_2\text{-O})] \cdot 0.5\text{C}_6\text{H}_6$ (**I**) and dimethylsulfoxide $[(\text{Ph}_3\text{Sb})_2(\mu_2\text{-O}_2\text{AsMe}_2)_2(\mu_2\text{-O})] \cdot 0.5\text{dmsO}$ (**II**). According to X-ray data, in complex **I** antimony atoms, linked together by two bidentate dimethylarsinato-ligands and a bridging oxygen atom have the configuration of an octahedron: axial angles CSbO are $167.62(11)^\circ$ - $173.28(10)^\circ$. The bond lengths of Sb-C are $2.156(3)$ - $2.176(3)$ Å. The distances of Sb-O(As) are $2.111(2)$ - $2.158(2)$ Å, in the fragment Sb-O-Sb bond lengths are $1.967(2)$ Å and $1.968(2)$ Å. Arsenic atoms have tetrahedral configuration (CAsC $108.03(17)^\circ$ - $110.29(16)^\circ$, CAsO $106.62(14)^\circ$ - $110.66(13)^\circ$). Bond lengths As-C and As-O are $1.911(3)$ - $1.915(3)$ Å and $1.681(2)$ - $1.691(2)$ Å respectively.