

Synthesis and structure of platinum complex $[\text{Ph}_4\text{Sb}(\text{dms})][\text{PtCl}_5(\text{dms})]$

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Abstract

The reaction of equimolar amounts of pentaphenylantimony with hexachloroplatinohydrogen acid in dimethylsulfoxide was conducted to synthesize the complex $[\text{Ph}_4\text{Sb}(\text{dms})]^+[\text{PtCl}_5(\text{dms})]^-$ (**I**). Reaction of chloride tetraphenylantimony with hexachloroplatinohydrogen acid in acetone or with hexachloroplatinate potassium in water lead to the formation of the complex $[\text{Ph}_4\text{Sb}]_2^+[\text{PtCl}_6]^{2-}$ (**II**), recrystallization of which from dimethylsulfoxide gives complex (**I**). According to the X-ray data in cations of complex **I** the atoms of antimony have trigonal-bipyramidal environment with the oxygen atom of the ligand in dimethylsulfoxide axial position ($\text{Sb}(1,2)\cdots\text{O}$ 2.541(2), 2.553(2) Å). In the octahedral anions dimethylsulfoxide ligand is coordinated with the platinum atom of sulfur atom ($\text{Pt}(1,2)\cdots\text{S}$ 2.305(2), 2.326(2) Å).