Full Paper

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Autocomplexes in salt melts of divalent metals

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Abstract

It has been shown that chemical equilibrium with respect to the formation or dissociation of the complex anions in salt melts of divalent metals halides is controlled by electrostatic interaction and excluded volume forces. The analysis of the equilibrium concentration of autocomplexes in simple statisticalthermodynamic model has been carried out. The possible types of chemical equilibrium shift towards dissociation were shown.