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Synthesis and crystal structure of tetrachloroaurate butyltriphenylphosphonium $[\text{Ph}_3\text{BuP}]^+[\text{AuBr}_4]^-$

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Abstract

By interaction of butyltriphenylphosphonium bromide with $\text{HAuBr}_4 \cdot 9\text{H}_2\text{O}$ in acetone we obtained the complex $[\text{Ph}_3\text{BuP}]^+[\text{AuBr}_4]^-$ (**I**). According to the data of X-ray, the complex **I** is presented by the crystalline tetrahedral cations $[\text{Ph}_3\text{BuP}]^+$ (bond lengths P-C_{Ph} and P-C_{Bu} 1.790(8)-1.799(8) and 1.790(8) Å, respectively, the angles CPC 107.9(4)°-110.7(4)°) and planar centrosymmetric anions $[\text{AuBr}_4]^-$ (angles BrAuBr 88.91(4)°-91.09(4)° and 177.64(5)°-180°, the bond lengths Au-Br are 2.4104(10)-2.4246(11) Å).