Thematic Section: Chemical and Technological Research. **Full Paper** Subsection: Organic Chemistry. Registration Code of Publication: 13-36-12-97 Publication is available for discussion in the framework of the on-line Internet conference "Butlerov readings". http://butlerov.com/readings/ Contributed: December 4, 2013.

Topic: Kinetics and mechanism of acyl transfer reactions. Part 6.

Quantum chemical interpretation of dipeptides and aminoacids reactivity in processes of acids amides and sulfamides formation

© Ludmila B. Kochetova,¹ Natalia V. Kalinina,² Tatiana P. Kustova,¹*⁺ and Lev V. Kuritsyn²

¹ Department of Organic and Physical Chemistry. Ivanovo State University. Ermak St., 39. Ivanovo, 153025. Russia. Phone: +7 (84932) 37-37-03. E-mail: kustova t@mail.ru

² Department of Inorganic and Analytical Chemistry. Ivanovo State University.

Ermak St., 39. Ivanovo, 153025. Russia. Phone: +7 (84932) 37-37-03.

*Supervising author; ⁺Corresponding author Keywords: acylation, aminoacids, dipeptides, benzovl chloride, sulfonyl chlorides, benzoic acid esters, quantum chemical calculation, reaction mechanism, potential energy surface.

Abstract

Quantum chemical simulation of aminoacids and dipeptides anions, as well as chloroanhydrides of benzoic, sulfobenzoic, 3-nitro-, 4-methylsulfobenzoic acids and nitro substituted phenyl esters of benzoic acids is carried out by ab initio methods. It is established that values of nitrogen atoms charges can be used as reactivity indexes of nucleophiles in N-acylation; acylation agents reactivity is connected with LUMO energies of their molecules. PES of glycine reaction with 4-methylbenzene sulfonyl chloride in gaseous phase is calculated, it is shown that the reaction occurs by $S_N 2$ mechanism.