

Quantum-chemical study of ozone protonation

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Abstract

The process of protonation of ozone in the singlet and triplet states was studied using B3LYP/6-311++G(df,p) quantum-chemical method. The geometric structure of the products of protonation were studied. The thermodynamic parameters of the reaction of protonation of ozone in the singlet and triplet electronic configurations were described. It was shown that the process of protonation of ozone in all cases is exothermic and proceeds with free energy and entropy decrease.