

Quantum chemical and experimental study of anionic σ -adducts of 2-methyl-5,7-dinitrobenzo[d]oxazole with methoxide ion

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Abstract

The charges on atoms and the contributions of p_z -orbitals of the atoms in LUMO in the molecules 2-methyl-5,7-dinitrobenzo[d]oxazole was calculated by semi-empirical method PM6 in the gas phase and in methanol. The standard heat of formation of the anticipated products of addition of the methoxide ion to the substrate, the standard enthalpy and Gibbs energy of reactions was calculated. It is found that in methanol the decisive role played by the charge factor, the attack of the nucleophile is most likely at the carbon atoms C-2, C-4 and C-6 of the substrate. The σ -adduct, which is a structure with the position of the methoxyl group at the second carbon atom of oxazole cycle, was selected under the action of cesium carbonate on 2-methyl-5,7-dinitrobenzo[d]oxazole in methanol at room temperature. The structure of the obtained compounds was proved by NMR (¹H NMR, ¹³C NMR, HSQC, HMBC), IR spectroscopy, and elemental analysis.