

## Mono- and polynuclear heteroligand complexonates of cobalt(II) in the presence of $\beta$ -hydroxy- $\alpha$ -aminopropionic acid

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### Abstract

Equilibria studies in the systems containing cobalt(II) ion, ethylenediaminetetraacetic and  $\beta$ -hydroxy- $\alpha$ -aminopropionic (serine) acids were performed with spectrophotometric and pH-metric methods. The stoichiometry of the complexes formed, their pH ranges of existence, stability constants and accumulation fraction were calculated depending on the concentration of ligands and acidity of solution. Experimental data were processed using mathematical models which allowed to establish the presence in the solution of complex particles of general composition  $[\text{ComSernEdtar}]^{2m-n-4r}$  ( $m = 1-3$ ,  $n = 0-6$ ,  $r = 0-1$ ). Presents the possible structures of the complexes formed.

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