

Thematic course: Chemical bath synthesis of metal chalcogenide films. Part 30.

Influence of reaction mixture ligand background at the formation of CdS – PbS thin films by chemical bath deposition

© Natalia A. Forostyanaya, Larisa N. Maskaeva, Anastasia D. Kutyavina,

Maria A. Ponomareva, Anastasia A. Rozhina, Polina O. Mihnevich and Vyacheslav F. Markov

¹Physical and Colloidal Chemistry Department. Federal State Autonomous Educational Institution of Higher education “Ural Federal University named after the first President of Russia B.N. Yeltsin”. Mira St., 19.

Yekaterinburg, 620002. Sverdlovsk Region. Russia. Phone: +7 (343) 375-93-18. E-mail: mln@ural.ru

²Department of Chemistry and Combustion Processes. Ural Institute of Fire Service of EMERCOM of Russia.

Mira St., 22. Ekaterinburg, 620022. Sverdlovsk Region. Russia. Phone: +7 (343) 360-81-68.

*Supervising author; [†]Corresponding author

Keywords: ionic equilibrium, ligands, hydrochemical deposition, thin films, lead sulfide, cadmium sulfide, substitutional solid solutions.

Abstract

The concentration planes of lead and cadmium sulphide formation in “ $\text{Cd}^{2+} - \text{Pb}^{2+} - \text{L} - \text{OH}^- - \text{N}_2\text{H}_4\text{CS}$ ” systems, where $\text{L} - \text{C}_6\text{H}_5\text{O}_7^{3-}$, $\text{H}_2\text{NCH}_2\text{CH}_2\text{NH}_2$, $\text{C}_6\text{H}_5\text{O}_7^{3-} + \text{H}_2\text{NCH}_2\text{CH}_2\text{NH}_2$, $(\text{HO}-\text{CH}_2\text{CH}_2)_3\text{N} + \text{H}_2\text{NCH}_2\text{CH}_2\text{NH}_2$, $(\text{HO}-\text{CH}_2\text{CH}_2)_3\text{N}$ were calculated by analyzing of the ion equilibrium. Scanning electron microscopy and energy dispersive analysis results showed the effect of the nature and strength of the complexing agents on composition and morphology of synthesized nanocrystalline CdS – PbS films.