

Some quantum considerations on oscillation characteristics of oxyhydrate gels

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Abstract

An ordinary nonlinear reaction-diffusion equation is known: $\frac{\partial}{\partial t} n = D\Delta n - (E - U)n \ln n$. Those equations describe chemical processes that occur in oxyhydrate colloid systems and are called equations with Liesegang operator. The state of a colloid is fully set by the function of a phase as a matrix colloid concentration. No multiplier $\ln n$ is used unless concentrations of the matrix forming gel change within considerable limits; in other words, if the influence of the logarithm is weak.

A correlation between the equation $\frac{\partial}{\partial t} n = D\Delta n - (E - U)n \ln n$ obtained and the previous equation $\frac{\partial}{\partial t} n = D\Delta n + L[n]$ was discovered, which means that the Liesegang operator is written down:

$$L[n] \equiv -(E - U)n \ln n,$$

as gel phase function dynamics: + sign in the operator corresponds to energy loss by this gel phase, while – sign – to energy accumulation. This is why entanglement expressed by the equation of Liesegang systems is a special quantum form of correlation between compound systems, which has no classical analog, and forms in a compound system that consists of two or more interacting systems (or those that had interacted and then were separated), and is a superposition of macroscopically quantized states, with fluctuations in individual parts being interrelated through non-local quantum correlations, not through classical interactions (correlations).