

Synthesis and structure of *bis*(tetraphenylantimony) oxalate dioxane solvate

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Abstract

The reaction of pentaphenylantimony and oxalic acid (2:1 mol, 2 h, 90°C) in dioxane has led to the formation of *bis*(tetraphenylantimony) oxalate dioxane solvate (**1**), the structure of which is determined by X-ray diffraction analysis. The antimony atoms of the binuclear compound **1** ($M = 1036.42$, monoclinic, $P2_1/c$, $a = 11.3204(4)$ Å, $b = 20.2822(7)$ Å, $c = 22.7088(7)$ Å, $\beta = 93.8490(10)^\circ$, $V = 5202.2(3)$ Å³, $Z = 4$, $\rho = 1.323$ g/cm³, $\mu = 1.083$ mm⁻¹, $F(000) = 2088.0$, $R_{\text{int}} = 0.0942$, $GOOF = 1.028$, $R_1 = 0.0570$, $wR_2 = 0.1761$) has distorted octahedral coordination (axial angles CSbC, OSbC are equal to $156.6(3)$ - $158.9(3)^\circ$, $161.6(2)$ - $165.0(2)^\circ$, respectively) with tetradentate bridging oxalate ligand (Sb–C 2.142(7)-2.339(5) Å). The distances Sb–O, O–C, C–C in two five-membered metallocycles [SbO₂C₂] equal 2.329(5)-2.357(4) Å, 1.239(8)-1.262(8) Å, 1.571(9) Å, respectively.