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Synthesis and structure of *bis*(tetraphenylantimony) oxalate dioxane solvate

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Abstract

The reaction of pentaphenylantimony and oxalic acid (2:1 mol, 2 h, 90°C) in dioxane has led to the formation of bis(tetraphenylantimony) oxalate dioxane solvate (1), the structure of which is determined by Xray diffraction analysis. The antimony atoms of the binuclear compound 1 (M = 1036.42, monoclinic, $P2_1/c$, a = 11.3204(4) Å, b = 20.2822(7) Å, c = 22.7088(7) Å, $\beta = 93.8490(10)^\circ$, V = 5202.2(3) Å³, Z = 4, $\rho = 1.323$ Γ/cm^3 , $\mu = 1.083 \text{ mm}^{-1}$, F(000) = 2088.0, $R_{\text{int}} = 0.0942$, GOOF = 1.028, $R_1 = 0.0570$, $wR_2 = 0.1761$) has distorted octahedral coordination (axial angles CSbC, OSbC are equal to 156.6(3)-158.9(3)°, 161.6(2)-165.0(2)°, respectively) with tetradentate bridging oxalate ligand (Sb-C 2.142(7)-2.339(5) Å). The distances Sb-O, O-C, C-C in two five-membered metallocycles [SbO₂C₂] equal 2.329(5)-2.357(4) Å, 1.239(8)-1.262(8) Å, 1.571(9) Å, respectively.