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## Thematic course: gem-Dichlorocyclopropylmethyl substituted nitrogen containing heterocycles. Part 3. Synthesis and properties of 3-[(1-methyl-2,2-dichlorocyclopropyl) methyl]hydantoins

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## Abstract

Hydantoin derivatives containing aromatic or heterocyclic substituents attract attention as drugs and antibacterial agents. Introduction of gem-dichlorocyclopropyl group to their composition causes the considerable interest because three-membered carbocyclic group is a nutrient and also has high reactivity. Therefore, an actual task is the development of methods of synthesis of *gem*-dichlorocyclopropylhydantoins.

In this report are given results on studying of reactions 1-methyl-1-chlormethyl-2,2-dichlorocyclopopan (1) with 5,5-dimethyl- (2a), 5,5-diphenyl- (2b) and 1-phenylhydantoins (2c) in the environment of dimethylformamide in the presence of sodium hydroxide. Effective use of cheap sodium hydroxide was confirmed by the example of 5,5-dimethylhydantoin alkylation chloride (1).

Carrying out these reactions, it was necessary to take into account that the alkylating reagent 1 comprises a different activity on halogen atoms, and hydantoins (2a-c) – imide group. This does not exclude the possibility of leakage in the presence of sodium hydroxide in a polar solvent competing reactions. We have found that when an equimolar ratio of the reactants and temperature of 120-125 °C in high yields the products are formed exclusively by gem-dichlorocyclopropyl in position "3" hydantoin cycle -1-R-5,5-R<sup>1</sup><sub>2</sub>-3-[(1-methyl-2,2-dichlorocyclopropyl)methyl]hydantoins [3a-c, R = H,  $R^1 = CH_3(a)$ , Ph (b); R = Ph,  $R^1 = H(c)$ ].

Obviously, in the above conditions initially hydantoins corresponding sodium salts (2a-c) are formed which are further alkylated by cyclopropane 1 according to halogenomethyl group without exposing cyclopropane ring.

Structure of compounds **3a-c** was confirmed by nuclear magnetic resonances methods <sup>1</sup>H NMR and IR spectroscopy. The IR spectrum contains signals charactering carbonyl group, and *gem*-dichlorocyclopropyl and benzene rings. In the analysis of <sup>1</sup>H NMR spectra we established magnetic nonequivalence methylene protons of the cyclopropane ring and CH<sub>2</sub>N-group.

Thus, reaction of 1-methyl-1-chloromethyl-2,2-dichlorocyclopropane 5-phenyl-, 5,5-diphenyl- and 5,5dimethylhydantoins in the presence of sodium hydroxide in dimethylformamide medium at equimolar ratio of reagents are received corresponding 3-[( gem- dichlorocyclopropyl)methyl]substituted hydantoins. We studied their physical and chemical properties. Synthesized 3-substituted hydantoins can potentially have biological activity.