Submitted on December 10, 2016.

Development of carbon composite based on carbon fibers from viscose material for supercapacitor electrodes

© Mikhail V. Astakhov, Anatoly T. Kalashnik, Ekaterina E. Kazenas, Tatiana L. Lepkova, Vladimir.V. Kozlov, and Farrukh S. Tabarov

Physical Chemistry Division. National University of Science and Technology MISiS. Leninsky Prospekt, 6. Moscow, 119049. Russia. Phone: +7 (495) 638-46-64. E-mail: fantotsi.0104@mail.ru

*Supervising author; ⁺Corresponding author

Keywords: carbon fiber, electric double layer, pseudo capacity, capacity, electrodes.

Abstract

Compact carbon material for supercapacitor electrodes (SC), was made from a carbon fiber (CF) derived from viscose materials. Samples of viscose material were impregnated with an aqueous solution of phosphoric acid and were dried for two hours and subjected to carbonization in an argon stream at a heating rate of 50 °C/min to a maximum temperature of 700 to 900 °C. After reaching the maximum temperature, the CF were kept at a constant temperature, and subjected for activation in the gas stream. After activation, the fiber was cooled to room temperature in an argon stream. Activity of adsorption was investigated by sorption of methylene blue indicator. The morphological features were studied by the scanning JMS-1700F Japanese company JEOL electron microscope. Assembling of electrode were used the CF (C = 80 % wt.), milling in a mortar, a binder which is a suspension of PTFE grade F4D in an amount of 10% by weight. To increase the conductivity of the material was added 10 % wt. of soot (CABOT® VULCAN® X C72). The mixture was rolled on rollers. This method yielded a tape used as a material for electrodes. The tape was pre-dried at 120 °C, a symmetrical supercapacitor cell samples were collected from the tape (SC), consisting of two electrodes and the separator. Conductive carbon was coated on an aluminum foil that served as a current collector. Electrochemical characteristics of SCs cells were examined by galvanostatic charge-discharge analyzer HIT ASK 2.5.10.8 and cyclic voltammetry on potentiostat JPC 2000. The salt of triethylmethylammonium tetrafluoroborate $(C_2H_3)_3CH_3NBF_4$ in acetonitrile served as an electrolyte in the cells of the SC. After activation of hydrocarbons in the CO2 stream is characterized by having a developed surface that capable to a high specific capacity of CF. By the value of electrochemical data test the specific capacitance of cells was 120 F/g (18.9 F/cm³) at current density 1 mA/cm². The variation of capacitance depending on the number of cycles indicates a stable capacitance value during the entire process of cycling of SC cells. Residual capacity after 2000 cycles decreased by 14% from the initial value, which indicates a good stability of structure of the electrode material of viscose.

References

- [1] E. Frackowiak. Carbon materials for supercapacitor application. *Phys. Chem. Chem. Phys.* 2007. Vol.9. P.1774-1785.
- [2] E. Frackowiak, F. Béguin. Carbon materials for the electrochemical storage of energy in capacitors. Carbon. 2001. Vol.39. P.937-950.
- [3] Y. Huang, J. Liang, Y. Chen. An overview of the applications of graphene-based materials in supercapacitors. Small. 2012. Vol.8. P.1805-1834.
- [4] V. Presser, M. Heon, Y. Gogotsi. Carbide-derived carbons-from porous networks to nanotubes and graphene. Adv. Funct. Mater. 2011. Vol.21. P.810-833.
- [5] E. Frackowiak, F. Béguin. Electrochemical storage of energy in carbon nanotubes and nanostructured carbons. Carbon. 2002. Vol.40. P.1775-1787.
- [6] G. Lota, K. Fic, E. Frackowiak. Carbon nanotubes and their composites in electrochemical applications. Energy Environ. Sci. 2011. Vol.4. P.1592-1605.,
- [7] Y.-G. Wang, H.-Q. Li, Y.-Y. Xia. Ordered Whiskerlike Polyaniline Grown on the Surface of Mesoporous Carbon and Its Electrochemical Capacitance Performance. Adv. Mater. 2006. Vol.18. P.2619-2623.

Full Paper M.V. Astakhov, A.T. Kalashnik, E.E. Kazenas, T.L. Lepkova, V.V. Kozlov, and F.S. Tabarov [8] Mi H., Zhang X., anS., Ye X., Yang S. Microwave-assisted synthesis and electrochemical capacitance of

- polyaniline/multi-wall carbon nanotubes composite. *Electrochem. Commun.* 2007. Vol.9. P.2859-2862.
- [9] K. Jurewicz, S. Delpeux, V. Bertagna, F. Béguin, E. Frackowiak. Supercapacitors from nanotubes/polypyrrole composites. Chem. Phys. Lett. 2001. Vol.347. P.36-40.
- [10] X. Li, X.F. Fang, R.Z. Pang, J.S. Li, X.Y. Sun, J.Y. Shen, W.Q. Han, L.J. Wang. Self-assembly of TiO2 nanoparticles around the pores of PES ultrafiltration membrane for migration organic fouling. J.Membr. Sci. 2014. Vol.467. P.226-235.
- V. Presser, M. Heon, Y. Gogotsi. Carbide-derived carbons-from porous networks to nanotubes and [11] graphene. Adv. Funct. Mater. 2011. Vol.21. P.810-833.
- [12] G. Salitra, A. Soffer, L.Eliad, Y. Cohen, D. Aurbach. Carbon Electrodes for Double-Laver Capacitors I. Relations Between Ion and Pore Dimensions. Journal of The Electrochemical Society. 2000. Vol.147/ No.7. P.2486-2493.
- [13] L. Eliad, G. Salitra, A. Soffer, D. Aurbach. Ion Sieving Effects in the Electrical Double Layer of Porous Carbon Electrodes: Estimating Effective Ion Size in Electrolytic Solutions. J. Phys. Chem. B. 2001. Vol.105. P.6880-6887.
- K.S. Sulaiman1,2 & A. Mat1,2 & A. K. Arof2 Activated carbon from coconut leaves for electrical [14] double-layer capacitor. Ionics. 2016. Vol.22. P.911-918.
- Li X, Xing W, Zhuo S, Zhou J, Li F, Oiao S-Z, Lu G-O (2011) Preparation of capacitor's electrode [15] from sunflower seed shell. Bioresour Technol. 102(2):1118-1123.
- [16] Elmouwahidi A, Zapata-Benabithe Z, Carrasco-Marín F, MorenoCastilla C (2012) Activated carbons from KOH-activation of argan (Argania spinosa) seed shells as supercapacitor electrodes. Bioresour Technol. 111:185-190.
- B.E. Conway. Electrochemical supercapacitors scientific fundamentals and technological [17] applications. New York: Kluwer Academic/Plenum. 1999.
- [18] B.E.Conway. Transition from 'supercapacitor' to 'battery' electrochemical capacitor applications. Electrochem Soc. 1991. Vol.138. No.6. P.1539-1548.
- [19] A.K. Arof, M.Z. Kufian, M.F. Syukur, M.F. Aziz, A.E. Abdelrahman, S.R. Majid. Electrical double layer capacitor using poly(methyl methacrylate)–C₄BO₈Li gel polymer electrolyte and carbonaceous material from shells of matakucing fruit. *Electrochim Acta*. 2012. Vol.74. P.39-45.