

Biomimetic composites based on carbonatation, albumin and gelatin for biomedical applications

© Svetlana A. Gerk,⁺ Olga A. Golovanova,* Irina M. Zyryanova, and Viktoria N. Odajiu

Inorganic Chemistry Division. F.M. Dostoevsky Omsk State University.

Mir Ave. 55a. Omsk, 644077. Russia. Phone: +7 (3812) 268-199.

E-mail: gerksa_11@mail.ru, golovanoa2000@mail.ru

*Supervising author, ⁺Corresponding author

Keywords: composites, biomaterials, hydroxyapatite, protein, albumin, gelatin, crystallization, thermal degradation, resorption, isomorphism.

Abstract

In the present work, biocomposites were obtained by the method of deposition from a prototype of a human biofluidic fluid with varying concentrations of albumin and gelatin. Materials synthesized from media with albumin and gelatin (<7 g/l) are represented by carbonate hydroxyapatite and biopolymer, and from solutions containing >7 g/l gelatin, include 10% by weight. It consists of crystallites of smaller size. It is shown that the presence of proteins in the model medium of less than 7 g/l does not affect the crystal-chemical parameters of the resulting aggregates. Such apatites are characterized by a mixed AB-type substitution of structural anions by carbonate groups. At maximum concentrations of biopolymers in solution (albumin – 12 g/l, gelatin – 10 g/l), the formation of predominantly A-type carbonate hydroxyapatite occurs. Thermochemical transformations of composites were studied. It is shown that the loss of mass of carbonate hydroxyapatite-albumin composites during annealing occurs due to an increase in the mass loss of protein and crystallization water and is independent of the albumin content in the model solution at ≤ 7 g/l. It is noted that the most heat-stable are gelatin-containing samples. The thermal stability of carbonate hydroxyapatite-gelatin composites decreases with increasing biopolymer content in the solid phase. It was found that the presence of proteins results in the deposition of composites with a lower degree of crystallinity and a specific surface area. The kinetics of dissolution of composites in neutral and weakly acidic media was studied. It is established that the resorption in 0.9% NaCl of biocomposites does not depend on the nature of the biopolymer, nor on its concentration in solution. Powders obtained from viscoelastic media dissolve at a higher rate in the acetate buffer solution. The most soluble in weakly acidic conditions are samples that precipitate from a medium containing more than 10 g/l albumin and 7 g/l gelatin.

References

- [1] V.K. Leontyev. Biologically active calcium phosphate-containing materials for dentistry. *Stomatology*. **1996**. No.5. P.4-6. (russian)
- [2] V.M. Bezrukov, A.S. Grigoryan. Hydroxyapatite as a substrate for osseous plastics: theoretical and practical aspects of the problem. *Stomatology*. **1996**. Vol.75. No.5. P.7-12. (russian)
- [3] T.V. Safronova, V.I. Putlyaev. Medical inorganic materials science in Russia: calcium phosphate materials. *Nanosystems: Physics, Chemistry, Mathematics*. **2013**. No.4(1). P.24-47. (russian)
- [4] V.N. Gorshenev, Yu.A. Ershov, A.T. Teleshev, E.D. Sklyanchuk, A.A. Prosvirin, S.A. Grigoriev. Hydroxyapatite composites for medical use. *Medical Equipment*. **2014**. No.1. P.30-33. (russian)
- [5] E. Bourgeat-Lami. Organic-inorganic nanostructured colloids. *Journal Nanoscience and Nanotechnology*. **2002**. Vol.2. №1. P.1-24.
- [6] A. Stein, B.J. Melde, R.C. Schroden. Hybrid inorganic-organic mesoporous silicates nanoscopic reactors coming of age. *Advanced Materials*. **2000**. Vol.12. No.19. P.1403-1419.
- [7] A.L. Boskey. Mineralization of Bones and Teeth. *Elements*. **2007**. Vol.3. P.387-393.
- [8] C. Rey, C. Combes, C. Drouet, M.J. Glimcher. Bone mineral: update on chemical composition and structure. *Osteoporosis International*. **2009**. No.20(6). P.1013-1021.
- [9] R. Murugan, S. Ramakrishna. Crystallographic study of hydroxyapatite bioceramics derived from various sources. *Crystal Growth and Design*. **2005**. Vol.5. P.111-112.

- [10] S.M. Barinov. Ceramic and composite materials based on calcium phosphates for medicine. *Successes in Chemistry*. **2010**. No.1. P.15-32. (russian)
- [11] A.P. Golovan, A.A. Rugal, V.M. Gunko, V.N. Barvinchenko, Ya. Skubishevskaya-Ziembra, R. Lebeda, T.V. Krupskaya, V.V. Turov. Modeling of bone tissue by composite systems based on hydroxyapatite - albumin - gelatin. *Surface*. **2010**. Iss.2(17). P.244-265. (russian)
- [12] A.A. Afonko, S.A. Kirilova, V.I. Almyashev. Ceramic and composite nanomaterials based on calcium orthophosphates. *Nanosystems: Physics, Chemistry, Mathematics*. **2012**. No.3(5). P.84-102. (russian)
- [13] Pat. 2516921. The Russian Federation. IPC A61K 35/32 (2006.01) Composition for bone plasty (variants). Larionov E.V., Ivanov S.Yu., Muraev A.A., Anisimov S.I. 2013115516/15, applic. 08.04.2013, publ. 05.20.2014. (russian)
- [14] Yu.A. Petrovich, A.N. Gurin, N.A. Gurin, S.M. Kichenko. Prospects for the use in the dentistry of polyfunctional biopolymers chitosan and alginate. *Russian Stomatology*. **2008**. No.2. P.66-73. (russian)
- [15] J. Sundaram, T.D. Durance, R. Wang. Porous scaffold of gelatine – starch with nanohydroxyapatite composite processed via novel microwave vacuum drying. *Acta Biomaterialia*. **2008**. Vol.4. P.932-942.
- [16] M. Chen, J. Tan, Y. Lian, D. Liu. Preparation of Gelatin coated hydroxyapatite nanorods and the stability of its aqueous colloidal. *Applied Surface Science*. **2008**. Vol.254. P.2730-2735.
- [17] M. Li, X. Xiao, R. Liu, C. Chen, L. Huang. Structural characterization of zinc-substituted hydroxyapatite prepared by hydrothermal method. *Journal of Materials Science: Materials in Medicine*. **2008**. No.19. P.797-103.
- [18] H.C. Kroese-Deutman, J. van den Dolder, P.H. Spauwen, J.A. Jansen. Influence of RGD-loaded titanium implants on bone formation in vivo. *Tissue Engineering*. **2005**. No.11. P.1867-1875.
- [19] Pat. 2496150. The Russian Federation. IPC G09B 23/28 (2006.01) / A method for modeling the bone mineralization of coxarthrosis in vitro. Golovanova O.A., Lemesheva (Gerk) S.A., Izmailov R.R. 2012115045/14, applic. 16.04.2012, publ. 10.20.2013. (russian)
- [20] E.L. Matveeva, E.S. Spirkina, A.G. Hasanova. Biochemical composition of the synovial fluid of the knee joint of people is normal. *Successes of Modern Natural Science*. **2015**. No.9 (1). P.122-125. (russian)
- [21] V.N. Pavlova. Synovial environment of the joints. *Moscow: Medicine*. **1980**. 296p. (russian)
- [22] V.V. Bazarny. Synovial fluid (clinical and diagnostic significance of laboratory analysis). *Ekaterinburg: UGMA*. **1999**. 62p. (russian)
- [23] S.A. Gerk, O.A. Golovanova, and V.A. Klushin. Phase, elemental, amino acid and structural composition of physiogenic minerals. *Butlerov Communications*. **2012**. Vol.32. No.12. P.80-89. DOI: jbc-02/12-32-12-80.
- [24] D.W. Green, T.K. Goto, K.S. Kim, H.S. Jung. Calcifying tissue regeneration via biomimetic materials chemistry. *Journal of the Royal Society Interface*. **2014**. No.11. P.1-11.
- [25] S.V. Tsibulya, S.V. Cherepanova. Introduction to the structural analysis of nanocrystals. *Novosibirsk: Novosibirsk State University*. **2008**. 92p. (russian)
- [26] D.L. Goloshchapov, V.M. Kashkarov, N.A. Rumyantseva, P.V. Seredin, A.S. Lenshin, B.L. Agapov, E.P. Domashevskaya. Preparation of nanocrystalline hydroxyapatite by chemical deposition using a biogenic source of calcium. *Condensed Media and Interphase Boundaries*. **2011**. Vol.13. No.4. P.427-441. (russian)
- [27] E.S. Klimashina. Synthesis, structure and properties of carbonate-substituted hydroxyapatites for the creation of resorbable materials. *Author's abstract dis. ... cand. chem. sciences. Moscow: Moscow State University*. **2011**. 23p. (russian)
- [28] R.R. Izmailov, O.A. Golovanova. Bioresorbability of a granular composite based on carbonate hydroxyapatite and gelatin in media with different pH values. *Bulletin of Omsk University*. **2015**. No.2. P.61-65. (russian)
- [29] E.A. Losev. Investigation of crystalline phases formed in the systems "glycine-carboxylic acid" and "serine-carboxylic acid". *Dis. ... cand. chem. sciences. Novosibirsk*. **2014**. 116p. (russian)
- [30] A.S. Fomin, S.M. Barinov, V.M. Ievlev, I.V. Fadeeva, V.S. Komlev, E.K. Belonogov, T.L. Turaeva. Nanosized hydroxyapatite synthesized by precipitation in a solution of gelatin. *Reports of the Academy of Sciences*. **2006**. Vol.411. No.3. P.348-351. (russian)
- [31] A.P. Solonenko, O.A. Golovanova. Composites based on calcium phosphate and gelatin, obtained by varying the pH of the medium. *Bulletin of Omsk University*. **2013**. No.4. P.152-158. (russian)
- [32] V.G. Badelin, E.Yu. Tyunina, I.N. Mezhevoi. Thermogravimetric study of amino acids and aliphatic peptides. *Liquid Crystals and Their Practical Use*. **2014**. Vol.14. No.3. P.43-52. (russian)
- [33] V.V. Starikova, S.O. Rudchenko. Optimization of the properties of a composite based on hydroxyapatite and chitosan by varying the composition and heat treatment regimes. *Bulletin of Kharkov University*. **2010**. Iss.14. P.35-39.

- [34] V.C. Chang. Organic-inorganic interaction and the growth mechanism of hydroxyapatite crystals in gelatin matrices between 37 and 80 °C. *Journal of Materials Science*. **2006**. No.17. P.387-396.
- [35] S.M. Barinov. Bioceramics based on calcium phosphates. *Moscow: Science*. **2005**. 204p.
- [36] R.A. Landzberg. Laboratory workshop on physical and colloid chemistry. *Petropavlovsk-Kamchatsky: Kamchatka State Technical University*. **2004**. 75p. (russian)