Thematic Section: Research into New Technologies.

Full Paper

Reference Object Identifier – ROI: jbc-02/17-52-10-111 Subsection: Metallurgy and Materials Science. Publication is available for discussion in the framework of the on-line Internet conference "Butlerov readings". http://butlerov.com/readings/ Submitted on October 23, 2017.

Hydro-, pyrometallurgical method of obtaining alloys system Fe-Ni-Cr-Mn-Si

© Boris D. Khalezov, Oleg V. Zayakin,*⁺ Alexey S. Gavrilov, and Vladimir I. Zhuchkov

Institute of Metallurgy of Ural Division of Russian Academy of Science. Amundsen St., 101. Ekaterinburg, 620016. Sverdlovsk Region. Russia. Phone: +7 (343) 23-29-139. E-mail: zferro@mail.ru

*Supervising author; ⁺Corresponding author

Keywords: hydrometallurgy, sodium hydroxide, precipitation, pyrometallurgy, nickel, chromium, manganese, ferroalloy.

Abstract

The paper presents an analytical review of methods for processing oxidized nickel ores (ONO). It is shown that in modern conditions, domestic deposits of the ONO are not exploited at full capacity and by a technologically obsolete and environmentally dangerous scheme of processing of ONO by reductionsulphiding mine melting for matte. In our country, electrolytic nickel containing about 99% Ni is mainly used for the smelting of high-quality nickel steels, the high price of which does not contribute to the development of production of nickel-containing steels. The decision of the problem of profitability of processing of domestic poor oxidized nickel ores is the creation of a new promising technology for their processing. As an alternative to the existing restorative-sulphiding mine smelting on matte in Russia, it is possible to consider a hydro-, pyrometallurgical method for obtaining alloys of the Fe-Ni-Cr-Mn-Si system.

Experimental method showed that by hydrolytic precipitation (sodium hydroxide) at pH = 6.5, Al_2O_3 is completely recovered in the precipitate, which after calcination at 700 °C forms a powder of commercially pure alumina. When the remaining solution is treated with sodium hydroxide at pH 9.5, more than 99% of nickel and cobalt oxides, 92% of MnO and 46% of MgO are precipitated. As a result of selective deposition of oxidized nickel ore elements with sodium hydroxide, an oxide concentrate was obtained, which after roasting at 700 ° C contained, by mass. %: 67 NiO; 3 CoO; 20 MnO; 9 MgO.

For the processing of nickel-containing concentrate, a pyrometallurgical method for melting complex ferroalloys of the Fe-Ni-Cr-Mn-Si system is proposed, using ferrous selenium chromium as the reducing agent. A scheme of a hydro-, pyrometallurgical method for the processing of oxidized nickel ores has been developed, including: crushing, heap leaching, hydrolytic precipitation to produce aluminum and nickel-cobalt concentrates, silicothermic smelting to obtain complex alloys containing. %: 59% Ni; 17% Cr; 12 Fe; 6.5 Mn; 2.7 Si; 2.6 Co, suitable for melting stainless steels.

References

- [1] O.V. Zayakin, V.I. Zhuchkov, V.A. Salina, A.A. Babenko. Production of nickel and chromiumcontaining ferroallovs from domestic raw materials. Sat. tr. XIV International Congress of steelmakers and metal producers. Moscow: JSC Metallurgical Plant Elektrostal. 2016. P.290-292. (russian)
- [2] A.S. Gavrilov, B.D. Khalezov, A.V. Radushev, S.A. Petrova, V.N. Vaulina, and A.V. Kharitonova. Percolation leaching of oxidized nickel ores. Butlerov Communications. 2017. Vol.49. No.2. P.102-109. ROI: jbc-02/17-49-2-102
- [3] V.N. Mashchenko, V.A. Kniss, V.A. Kobelev, A.S. Avdeev. Preparation of oxidized nickel ores for smelting. Ekaterinburg: UB RAS. 2005. 321p. (russian)
- [4] V.V. Kozlova. Mineral resources of foreign countries: Nickel. *Moscow: VNII.* 1994. 192p. (russian)
- [5] V.A. Generalov, I.D. Reznik, T.A. Kharlakova. Methods for obtaining ferronickel from oxidized nickel ores (Part I). Non-ferrous metals. 1995. No.5. P.13-17. (russian)
- [6] I.D. Reznik, G.P. Ermakov, J.M. Shneerson. Nickel: in 3 vol. Oxidized nickel ores. Characteristics of ores. Pyrometallurgy and hydrometallurgy of oxidized nickel ores. Moscow: Science and Technology. 2004. Vol.2. 468p. (russian)
- [7] F. Ntuli, A.E. Lewis. Kinetic modeling of nickel powder precipitation by high-pressure hydrogen reduction. Chemical Engineering Science. 2009. Vol.64. P.2202-2215.

Full Paper

- [8] K. Wang, J. Li, R.G. McDonald, R.E. Browner. The effect of iron precipitation upon nickel losses from synthetic atmospheric nickel laterite leach solutions: Statistical analysis and modelling. Hydrometallurgy. 2011. Vol.109. P.140-152.
- [9] O. Coto, F. Galizia, I. Hernandez, J. Marrero, E. Donati, Cobalt and nickel recoveries from tailings by organic and inorganic bio-acids. Hydrometallurgy. 2008. Vol.94. P.18-22.
- [10] M.I. Gasik, N.P. Lakishev. Theory and technology of electrometallurgy of ferroalloys. *Moscow: JV* Intermet Engineering. 1999. 764 p. (russian)
- [11] V.I. Fadeev, Ya.I. Ostrovsky, A.A. Zernikel and others. Resource-saving technology for the production of low-carbon ferrochrome with the production of stabilized slags. Sat. reports: Modern resource-saving technologies. Odessa: ONU them. I.I. Mechnikov. 2012. P.131-134. (russian)
- [12] V.I. Zhuchkov, L.A. Smirnov, V.P. Zayko, Yu.I. Voronov. Technology of manganese ferroalloys. Part 2. Low-carbon ferromanganese. Ekaterinburg: Izd. UB RAS. 2007. 442p. (russian)