

Application of continuous flow analysis for determination of exchangeable magnesium in the basic types of Russian soils

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Keywords: Exchangeable magnesium, segmented flow analyzer, soils, methods of analysis.

Abstract

It is not possible to obtain a high yield of crops of good quality in the absence of balanced nutrition including sufficient supply of magnesium. Such imbalance also is a bad influence on the efficiency of fertilizers use.

Magnesium is a multifunctional element of the plants nutrition. This element is included in the molecules of chlorophyll, pectines, phytin, cells membranes. It takes a part in synthesis of aminoacids, proteins, and more than 300 ferments. Magnesium also takes a part in biochemical reactions during the cellular respiration and increases plant resistance to unfavorable growing conditions. Unlike for other macronutrients, magnesium deficiency doesn't lead to the death of plants but it causes a decrease of crops properties and an increase of yield losses during the storage period. Magnesium is a low-renewable because of its high capability to migration along the soil profile and its removal with the harvest. Thus the constant soil magnesium supply control is of great importance for high quality of crops and for the ecologically and economically proved application of magnesium fertilizers. At present, to determine the content of exchangeable magnesium in soils there are a few methods widely used – photometry, atomic absorption spectroscopy and chelatometry according to the standard method GOST 26487-85. There are automated analytical systems, such as continuous flow analyzer San⁺. The metrological characteristics of such system are not inferior to the modern analytical instruments, such as atomic absorption and atomic emission spectrometry, without, in most cases, the negative impact of interference cations and anions present in the soil and plant extracts. The aim of this research was to investigate the opportunity of applying of the classical method modification to the flow analysis system.

As our findings demonstrate modification we proposed exceeds the conventional technique, possessing at the same time high automation potential. The values resulting from the analysis of reference samples were within the error margins of a standardized method (GOST 26487-85).

As our research has shown the modified method is can be considered as promising in terms of applicability for practice of agrochemical soil analysis in Russia for purposes of exchangeable magnesium determination. Capability to apply method to the instrument for continuous flow analysis has significant meaning in relation of the productivity rate rise of routine laboratory practice.

References

- [1] C. Sprengel. Die Lehre vom Dünger oder Beschreibung aller bei der Landwirtschaft gebräuchlicher vegetabilischer, animalischer und mineralischer Düngermaterialien, nebst Erklärung ihrer Wirkungsart. Leipzig. 1839.
- [2] M. Senbayram, A. Gransee, V. Wahle, H. Thiel. Role of magnesium fertilizers in agriculture: plant-soil continuum. *Crop Pasture Sci.* **2015**. Vol.66. P.1219-1229.
- [3] M.E. Maguire, J.A. Cowan. Magnesium chemistry and biochemistry. *Biometals*. **2002**. Vol.15. P.203-210.
- [4] Firsov S.S. Validation of magnesium fertilizers application efficiency. *Sostoyanie i perspektivnyi razvitiya nauchnyih issledovaniy v sotsialnoy, ekonomicheskoy i pravovoy sfere Sbornik nauchnyih statey po materialam mezhdunarodnoy nauchno-prakticheskoy konferentsii* [The current situation and prospects of

- scientific research development in social, economic and legal fields of study. Collection of scientific articles based on the International Scientific and Practical Conference materials]. **2015**. P.313-318. (russian)
- [5] W. Guoa, H. Nazimc. Magnesium deficiency in plants: An urgent problem. *The Crop Journal*. **2016**. Vol.4. P.483-91.
 - [6] A.N. Aristarkhov. Agrochemical reasons of magnesium fertilizers application, *Plodorodie* [Fertility]. **2002**. No.3. P.15-17.
 - [7] H.A. Mills, Jones Jr., J.B. Plant Analysis Handbook II. *MicroMacro Publishing, Athens*. **1996**. GA.
 - [8] J.M. Penalosa, M.D. Cáceras, M.J. Sarro. Nutrition of bean plants in sandculture: influence of calcium/potassium ratio in the nutrient solution. *J. Plant Nutr.* **1995**. Vol.18. P.2023-2032.
 - [9] V.Ya. Tikhomirova. Influence of soil properties, fertilizers, lime and weather conditions on the magnesium sufficiency of crops. *Agrokhimiya* [Agrochemistry]. **2011**. No.5. P.84-89. (russian)
 - [10] V.A. Litvinskiy, E.A. Grishina, V.V. Nosikov, S.L. Belopukhov. Continuous flow analyzers: Application of automatization approach from ecological studies to the classic methods of agrochemistry soil analyses. *Vestnik Kazanskogo Tekhnologicheskogo Universiteta*. **2016**. Vol.19. No.21. P.178-180. (russian)
 - [11] E.A. Grishina, V.A. Litvinskiy, V.V. Nosikov, S.L. Belopukhov. Chemical analysis of the mobile boron compounds using the continuous flow analyzer San⁺⁺ *Vestnik Kazanskogo Tekhnologicheskogo Universiteta*. **2016**. Vol.19. No.23. P.149-153. (russian)
 - [12] J.M. Kolthoff. Eine neue spezifische Farbreaktion auf Magnesium und eine einfache kolorimetrische Methode zur quantitative Bestimmung von Spuren dieses, Elements. *Biochem.* **1927**. Vol.2. No.185. P.344-348.
 - [13] H.J.G. Challis, D.F. Wood. Absorptiometric determination of magnesium in titanium and its alloys. *Analyst*. **1954**. Vol.79. P.762-770.
 - [14] A. Young, T.R. Sweet, B.B. Baker. Simultaneous Spectrophotometric Determination of Calcium and Magnesium. *Anal. Chem.* **1955**. Vol.27. P.356-359.
 - [15] H.J. Gitelman, C. Hurt, L. Lutwak. An automated spectrophotometric method for magnesium analysis. *Anal. Biochem.* **1966**. Vol.14. P.106.
 - [16] E. Gindler, et al. Colorimetric determination with bound "calmagite" of magnesium in human blood serum. *Clin. Chem.* **1971**. Vol.17. P.662.
 - [17] N. Peerzada, E. Kozlik. Spectrophotometric Determination of Magnesium with 2,2':6',2'' Terpyridine. *Analytical letters*. **1990**. Vol.23. No.6. P.1087-1093.
 - [18] Z. Tesfaldet, J. Van Staden, I. Stefan. Spectrophotometric determination of magnesium in pharmaceutical preparations by cost-effective sequential injection analysis. *Talanta*. **2004**. Vol.64. P.981-988.
 - [19] C.K. Mann, J.H. Yoe. Spectrophotometric Determination of Magnesium with Sodium 1-Azo-2-hydroxy-3-(2,4-dimethylcarboxanilido)-naphthalene-1'-(2-hydroxybenzene-5-sulfonate). *Anal. Chem.* **1956**. Vol.28. No.2. P.202-205.
 - [20] W.T. Elwell, and D.F. Wood. Analysis of the new metals. Titanium, zirconium, hafnium, niobium, tantalum, tungsten and their alloys. *Pergamon Press*. **1966**. P.286.
 - [21] C.K. Mann, and J.H. Yoe. *Anal. chim. Acta*. **1957**. Vol.16. P.155-160.
 - [22] State Standard 26487-85 Soils. Determination of exchangeable calcium and exchangeable (mobile) magnesium by CINAO methods (with amendment). Standards Publ., 1985. 13 p. (In Russian)
 - [23] E.E. Ludwig, and C.R. Johnson. Spectrophotometric Determination of Magnesium by Titan Yellow. *Ind. Eng. Chem. Anal. Ed.* **1942**. Vol.14. No.11. P.895-897.
 - [24] W.S. Gillam. A Photometric Method for the Determination of Magnesium. *Ind. Eng. Chem., Anal. Ed.* **1941**. Vol.13. P.499.
 - [25] D.W. Neill, and R.A. Neely. The estimation of magnesium in serum using titan yellow. *J. Clin. Pathol.* **1956**. Vol.9. No.2. P.162-163.
 - [26] A.H. Cornfield, and A.G. Pollard. Use of titan yellow for the determination of magnesium in plant material. *Journal of the Science of Food and Agriculture*. **1950**. Vol.1. P.357-358.
 - [27] R.W. Schmid, and C.N. Reilley. New complexone for titration of calcium in the presence of magnesium. *Anal. Chem.* **1957**. Vol.29. P.264-268.
 - [28] G. Schwarzenbach, W. Biedermann, and F. Bangerter, V.I. Komplexone. Neue einfache Titriermethoden zur Bestimmung der Wasserhärte. *Helv. chim. Acta*. **1946**. Vol.29. P.811-818.
 - [29] B.A. Levine, R.J.P. Williams, G.M.M. Scott. Cellular Function of Calcium in Plants. *CRC Press*. **1985**. 52p.
 - [30] E.G. Lamkin, and M.B. Williams. Spectrophotometric determination of calcium and magnesium in blood serum with arsenazo and EGTA. *Anal. Chem.* **1965**. Vol.37. P.1029-1031.

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- [31] I. Walinga, van der Lee, J.J., Houba, V.J.G., van Vark, W., Novozamsky, I. Plant Analysis Manual. **1995**. 257p.
- [32] M.H. Sorour, H.A. Hani, H.F. Shaalan, and Mayyada El-Sayed, M.H. Experimental screening of some chelating agents for calcium and magnesium removal from saline solutions. *Desalination and Water Treatment*. **2016**. Vol.57. Iss.48-49.
- [33] State Standard 26483-85 Soils. Preparation of salt extract and determination of its pH by CINAO method. *Standards Publ.* **1985**. 6p. (russian)