- [1] Macromolecular protein complexes. Structure and function. Ed. by J. Robin Harris, J. Marles-Wright. Switzerland: Springer. 2017. 572p.
- [2] Aggregation of therapeutic proteins. Ed. by W. Wang, Ch.J. Roberts. New-York: Wiley. 2010. 484p.
- [3] M. Ashrafuzzaman, J. Tuszynski Membrane Biophysics. Berlin, Heidelberg: Springer-Verlag. 2012. 178p.
- [4] V.A. Kabanov. Polyelectrolyte complexes in solution and in the condensed phase. *Successes Chemistry*. 2005. Vol.74. No.1. P.5-24. (russian)
- [5] Polyelectrolytes. Ed. by P.M. Visakh, O. Bayraktar, G.A. Picó. Switzerland: Springer. 2014. 388p.
- [6] Polyelectrolyte complexes in the dispersed and solid state. I Principles and theory. Ed. by M. Müller. Berlin, Heidelberg: Springer. 2014. 229p.
- [7] V.A. Izumrudov. The phenomena of self-assembly and molecular "recognition" in solutions of (bio) polyelectrolyte complexes. Successes Chemistry. 2008. Vol.77. No.4. P.401-414. (russian)
- [8] Polyelectrolyte complexes in the dispersed and solid state. II Application aspects. Ed. by M. Müller. Berlin, Heidelberg: Springer. 2014. 264p.

© Natalia N. Smirnova,^{1*+} Elena S. Ilina,² and Kirill V. Smirnov³

¹Department of Chemistry. Vladimir State University named after A.G. and N.G. Stoletovs. Gor'kogo St., 87. Vladimir, 600000. Russia. Phone: +7 (9422) 47-98-67. E-mail: smirnovann@list.ru ²OOO NPF «Adgesiv». B. Nijegorodskaya St., 77. Vladimir, 600016. Russia. *Phone:* +7 (4922) 47-55-55 *E*-mail: elena-ilina@bk.ru ³Department of Biology and Ecology. Vladimir State University named after A.G. and N.G. Stoletovs. Gor'kogo St., 87. Vladimir, 600000. Russia. Phone: +7 (9422) 47-97-53. E-mail: kirillv.smirnov@vandex.ru

*Supervising author; ⁺Corresponding author Keywords: bovine serum albumin, natural polyelectrolytes, carboxymethylcellulose, interpolyelectrolyte reactions, protein-polyelectrolyte complexes.

Abstract The interaction of bovine serum albumin in aqueous solution with sodium salt of carbo-

xymethylcellulose with different molecular mass was studied. It is shown that, protein-polyelectrolyte complexes (PPC) forms because of macromolecular reactions, which are stabilized mainly by electrostatic forces. To characterize the PPC composition the ϕ parameter used, which is defines as the ratio of concentration of ionic groups of polyelectrolyte per mole of protein molecules. Using spectrophotometry is was established that, in the studied system when components are mixed under optimal conditions (ratio of components, pH of the solution) complexes are formed, the composition of which corresponds to $\varphi \sim 60$ ([carboxymethylcellulose - CMC]/[bovine serum albumin - BSA] = 0.2-0.25 g/g). The degree of conversation in reactions of protein with polyelectrolyte is close to 0.9. The region of existence of PPC established. The maximum yield of the product of interpolyelectrolyte reaction is fixed at pH \leq 4 (ζ potential of BSA molecules is above +12 mV). The increase in the molecular mass of polyelectolyte shifts the point of beginning of intensive complexation into the area of high pH values, thus, increasing the diapason of interaction of components. The molecular mass of polymer electrolyte in protein-polyelectolyte systems should be considered as a factor that has a decisive influence on the structure of forming particles. The size of the forming complex particles depending on the PPC composition and on the molecular of polymer polyelectrolyte varies from 10 nm to 5.0 µm. The particles of micron size forms for CMC samples with molecular mass of $2.5-4.5 \cdot 10^5$ at a mass ratio of components in PPC 0.1, while the decrease in the molecular mass of polymer polyelectrolyte to 3.1 10⁴ requires an increase of this value to 0.2. Short-chained polyelectrolytes are advisable to use to stabilize the protein macromolecules.

Full Paper

Submitted on May 21, 2018.

Thematic course: Protein-polyelectrolyte complexes. Part 2. Complexes bovine serum albumin with carboxymethylcellulose. Effect of molecular weight of polyelectrolyte.

References

Thematic Section: Biochemical Research.

Subsection: Molecular Biology.

COMPLEXES BOVINE SERUM ALBUMIN WITH CARBOXYMETHYLCELLULOSE. EFFECT OF MOLECULAR... 142-147

- [9] C.L. Cooper, P.L. Dubin, A.B. Kayitmazer, S. Turksen. Polyelectrolyte-protein complexes. Current Opinion Coll. Int. Sci. 2005. Vol.10. P.52-78.
- [10] V. Boeris, B. Farruggia, B. Nerli, D. Romanini, G. Picó. Protein-flexible chain polymer interactions to explain protein partition in aqueous two-phase systems and the protein- polyelectrolyte complexe formation. Int. J. Biol. Macromol. 2007. Vol.41. P.286-294.
- [11] J. Lombardi, V. Woitovich, G. Picó, V. Boeris. Obtainment of a highly concentrated pancreatic serine proteases extract from bovine pancreas by precipitation with polyacrylate. Sep. Purif. Technol. 2013. Vol.116. P.170-178.
- [12] N. Valetti, M. Brassesco, G. Picó. Polyelectrolyte-protein complexes: a viable platform in the downstream processes of industrial enzymes at scaling up level. J. Chem. Technol. Biotechnol. 2016. DOI: 10.1002/jctb.5050.
- [13] V. Boeris, D. Spelzini, J. Salgado, G. Picó, D. Romanini, B. Farruggia. Chymotrypsin-polv vinvl sulfonate interaction studied by dynamic light scattering and turbidimetric approaches. Biochim. Biophys. Acta. 2008. Vol.1780. P.1032-1039.
- [14] A.N. Cherkasov. Rapid analysis of ultrafiltration. Membr. Struct. Separ. Sci. and Technol. 2005. Vol.40. No.14. P.2775-2801.
- [15] H.-D. Jakubke, H. Jeschkeit. Aminosäuren, Peptide, Proteine. Berlin: Akademie-Verlag. 1982. 457p.
- [16] M. Othman, A. Aschi, A. Gharbi. Polyacrylic acids-bovine serum albumin complexation: Structure and dynamics. Mat. Sci. Eng. 2015. DOI: 10.1016/j.msec.2015.08.057.
- [17] V.B. Skobeleva, A.V. Zinchenko, V.B. Rogacheva, A.B. Zezin. The interaction of weakly cross-linked polyamine with bovine serum albumin. Bulletin of Moscow University. Ser.2. Chemistry. 1998. Vol.39. No.4. P.268-271. (russian)
- [18] L. Ahmed, J. Xia, P. Dubin, E. Kokufuta. Stoichiometry and mechanism of complex formation proteinpolyelectrolyte coacervation. J. Macromol. Sci. Part A: Pure Appl. Chem. 1994. Vol.31. No.1. P.17-29.