

Study of a composition of anionic and cationic surfactants as a collector for flotation separation of nepheline

© Andrey I. Gorodov⁺, and Nikolay A. Shapovalov*

Department of Theoretical and Applied Chemistry. Belgorod State Technological University

Named after V. G. Shukhov. Kostyukova St., 46. Belgorod, 308012. Russia.

Phone: +7 (951) 133-42-73. E-mail: gorodov-andrey@mail.ru

*Supervising author; ⁺Corresponding author

Keywords: surfactants; composition of anionic and cationic surfactants; flotation; collector; nepheline; aluminum; tailing; nepheline concentrate.

Abstract

The results of studies of aluminum – containing raw materials-waste of apatite flotation of the enrichment plant of *Apatit JSC*. On the basis of the conducted researches the assumption of possibility of uses as the collector for the reverse flotation of nepheline the mix consisting of anionic and cationic surfactants. Conducted study colloid-chemical properties of anionic and cationic surfactants and their composition, the results of which proved the optimal ratio of components in the mix consisting. To confirm the possibility of using the mix consisting of anionic and cationic surfactants as a collector, laboratory tests on the reverse flotation of nepheline were carried out. During the tests, the composition of the mix consisting and the optimal consumption of reagents were established. The most effective as the collector in flotation of nepheline ores, was the mix consisting of anionic and cationic surfactants. The recommended flow rate of the collector is 500-600 gram/ton. As a result of flotation enrichment of the raw materials with a total aluminum oxide content of 23.8%, using recycled water, nepheline concentrate with an Al₂O₃ content of 28.27% was obtained. The concentrate yield is 62% of the initial load mass.

References

- [1] A.V. Akimova et al. State report on the status and use of mineral resources of the Russian Federation in 2013. *Ministry of natural resources and environment of the Russian Federation. Moscow. 2014.* 387p. (russian)
- [2] N.A. Shapovalov, A.A. Krayniy, A.I. Gorodov, I.S. Makuschenko. Study of the influence of different kinds of collectors and depressant flotation iron-containing minerals Mikhailovsky deposit. *Fundamental research. 2014.* Vol.9. P.318-323. (russian)
- [3] T.N. Muhina. Improving the efficiency of reverse flotation of nepheline using high molecular weight alkyl benzene sulfonates. PhD Theses on the Tech. Sciences: 25.00.13. *SPb. 2004.* 20p. (russian)
- [4] V.N. Lygach, G.V. Ladygina, Ju.E. Bryljakov, M.A. Kostrova. To increase the efficiency of nepheline production at ANOF-II Apatit OJSC by improving the reagent regimen of reverse nepheline flotation. *Mining information and analytical bulletin. 2007.* No.10. P.365-369. (russian)
- [5] N.A. Shapovalov, V.A. Poluektova, A.I. Gorodov, A.A. Krayniy, I.L. Vintskovskaya, M.M. Ryadinskiy. Domestic phosphorus surfactant – active collectors complex separation of apatite-nepheline dirt. *Fundamental research. 2015.* Vol.2. P.1689-1693. (russian)
- [6] V.A. Chanturiya, T.V. Chekushina. Progressive methods of enrichment and complex processing of natural and technogenic mineral raw materials (Plaksinsky readings – 2014). *Materials of the International meeting (16-19 September, 2014). Almaty. 2014.* 624p. (russian)
- [7] N.A. Shapovalov, L.H. Zagorodnyuk, A.I. Gorodov, A.Y. Shekina, A.A. Krainiy. Comprehensive use of waste after beneficiation of Magnetite fraction by floatation. *Actual Issues of Mechanical Engineering (AIME 2017). Advances in Engineering Research. 2017.* Vol.133. P.766-772.
- [8] Flotation reagent. URL: <http://inhibitor.ru/index.php?id=54> (date: 13.08.2018).
- [9] V. Tomasic, I. Stefanic, N. Filipovic-Vincekovic. Adsorption, association and precipitation in hexadecyltrimethylammonium bromide/sodium dodecyl sulfate mixtures. *Colloid and Polymer Science. 1999.* Vol.277. P.153-163.
- [10] K. Tsuchiya, J. Ishikake, T.S. Kim, T. Ohkubo, H. Sakai, M. Abe. Phase behavior of mixed solution of a glycerin-modified cationic surfactant and an anionic surfactant. *J Colloid Interface Science. 2007.* Vol.312. P.139-145.