

## Viscosity of the B<sub>2</sub>O<sub>3</sub>-CaO-NiO (FeO) systems melts

© Alexander S. Vusikhis,<sup>+</sup> Leopold I. Leontiev, Evgeny N. Selivanov,\*  
Viktor P. Chentsov, and Valery V. Ryabov

Institute of Metallurgy UB RAS. Amundsen St., 101. Yekaterinburg, 620016. Russia. E-mail: vas58@mail.ru

\*Supervising author; <sup>+</sup>Corresponding author

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### Abstract

Structure of the B<sub>2</sub>O<sub>3</sub> melt was analyzed as well as CaO modifier additives (25, 34, and 45 mass %) effect to it. As was proved, non-ring groups of associated rings proper to pure B<sub>2</sub>O<sub>3</sub> are transformed into BO<sub>2</sub>O – metaborate triangles. The released oxide ions increase the coordination number of modifying ions, which occupy cationic vacancies place in the most disordered part of the melt greed.

The B<sub>2</sub>O<sub>3</sub>-CaO-NiO and B<sub>2</sub>O<sub>3</sub>-CaO-FeO melts viscosity were measured by the method of vibration viscometry. Ratio of Boron to Calcium oxide mass fractions was taken as 3/1 and content of Nickel and Iron oxides in the range up to 5 and 20 % mass respectively. The experiments had been carried out using vibration viscometer operating in the mode of forced oscillations. The melt temperature was measured by Platinum – Platinum – Rhodium thermocouple. The measurements were carried out in cooling mode of the melt from 1800 K at 7-10 K/min speed. The viscosity temperature dependencies as well as its dependence on Nickel and Iron oxides were determined. Data processing was performed using the Table curve application software. Viscosity experimental data of the B<sub>2</sub>O<sub>3</sub>-CaO-NiO system melts for 1373, 1423, 1473, 1523, 1573 K temperatures have been described by the equation:  $\eta = a + b \cdot \exp(-cx)$ , and for B<sub>2</sub>O<sub>3</sub> – CaO – FeO melts by equation:  $\eta = a + b \cdot x^2 + c \cdot \exp(x) + d \cdot \exp(-x)$ . Experimental data and calculations results show good convergence.

The results are supposed to be used for describing the kinetics of metal reduction in bubbling processes, accompanied by the concentrations change of the oxides under reduction. The obtained information is useful for correction the slag melts properties in non-ferrous metal production.

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