

The study of the modification methylphenylsiloxane resin with dimethylene links metal compounds and boric acid

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Abstract

In this paper we study the effect of modified metal compounds and boric acid (E), where Element = B, Ti, Zr, Al on the thermal stability of methylphenylsiloxane resins derived from alkoxyasilanes.

Currently interested in the reaction for producing oligosilsesquioxanes based on acidolysis alkoxyasilanes. In this regard, the author was tasked with obtaining new methylphenylsiloxane resins with different properties.

In this paper, we study the properties of new methylphenylsiloxane resins (MFSS), modified metal compounds and boric acid. New MFSS obtained by a new universal technology-acidolysis mixture of methyltriethoxysilane (MTEOS) and phenyltriethoxysilane (PHTEOS) with various radicals, which are environmentally friendly raw materials. The obtained MFSS were characterized by NMR spectroscopy on ¹H and ²⁹Si nuclei. The spectra were recorded at room temperature in deuterioacetone using a Bruker AM-360 Fourier spectrometer. ²⁹Si NMR spectra were measured using the pulse program "Inverse Gated Heteronuclear Decoupling".

Thermogravimetric analysis was performed on the device Derivatograph-H (firm Mom). TGA studies were carried out in the argon atmosphere and in air at a heating rate of 10 °C/min.

The acidolysis reaction of methyltriethoxysilane and phenyltriethoxysilane is a convenient and versatile method for the synthesis of new heat-resistant methylphenylsiloxane resins. In the course of the study, it was found that the resins obtained on the basis of organoalkoxyasilanes are characterized by higher thermal and thermo-oxidative stability. It is shown that the modification methylphenylsiloxane resin dimethylsiloxane links in the main chain elementlocalname fragments of ≡Si-O-Element-O-Si≡, where element = B, Ti, Zr, Al increases its resistance to the level methylphenylsiloxane resin without dimethylsiloxane links.

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