

Instability of molecular structure of non-IPR isomer 17984 (C₁) of the C₇₆ fullerene and probable ways of its stabilization

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Abstract

It is well-known that non-IPR fullerenes are highly unstable. For this reason, they cannot be obtained as pristine fullerenes; however, some of them become stable as derivatives (exohedral or endohedral). In this article, we attempted to elucidate in detail molecular structure for such a non-IPR fullerene. Using theoretical approach supported by DFT calculations, the features of molecular structure of isomer 17894 (C₁) of fullerene C₇₆ with data about distribution of single, double and delocalized π -bonds as well its structural formula has been determined for the first time. The instability of the studied fullerene molecule caused by its open-shell structure and significant local overstrains related to the high folding angle value of pentagons in pentalene fragment. The supposed synthesis of the endohedral molecule starts with the ionic pair formation, i.e. anionic fragment of fullerene cage and metal cation electrostatically bound with it. It would lead to closing of open electron shell of fullerene and local overstrain release at pentalene fragment. As to the exohedral derivatives the probable positions of addends are discussed. Both methods in their own demonstrate the possibilities to stabilize the molecule of the C₇₆ isomer 17894. The elucidation and analysis of structural features along with electronic characteristics of non-IPR fullerene molecules appear to be useful for predicting the possibility of their synthesis as derivatives and will assist in determination of their reactivity. This will ensure the targeted production of fullerenes and their derivatives for the needs of medicine, electronics and other industries. The fundamental knowledge of the properties of nanoobjects, namely fullerenes, is actually developing as the independent direction with a long-term perspective.

References

- [1] P.W. Fowler, D.E. Manolopoulos. An Atlas of Fullerenes. *Dover Publ., Mineola N.Y.* **2006**. 392p.
- [2] R. Guan, M. Chen, F. Jin, S. Yang. Strain release of fused pentagons in fullerene cages by chemical functionalization. *Angew. Chem. Int. Ed.* **2020**. Vol.59. P.1048-1073.
- [3] W. Krätschmer, L.D. Lamb, K. Fostiropoulos, D.R. Huffman. Solid C₆₀: a new form of carbon. *Nature*. **1990**. Vol.347. No.6271. P.354-358.
- [4] A.R. Khamatgalimov, V.I. Kovalenko. Stability of Isolated-Pentagon-Rule isomers of fullerene C₇₆. *Fuller. Nanotub. Car. Nanostruct.* **2015**. Vol.23. P.148-152.
- [5] S. Yang, A.A. Popov, L. Dunsch. The role of an asymmetric nitride cluster on a fullerene cage: The non-IPR endohedral DySc₂N@C₇₆. *J. Phys. Chem. B.* **2007**. Vol.111. P.13659-13663.
- [6] M. Suzuki, N. Mizorogi, T. Yang, F. Uhlík, Z. Slanina, X. Zhao, M. Yamada, Y. Maeda, T. Hasegawa, S. Nagase, X. Lu, T. Akasaka. La₂@C_s(17490)-C₇₆: a new non-IPR dimetallic metallofullerene featuring unexpectedly weak metal-pentalene interactions. *Chem. Eur. J.* **2013**. Vol.19. P.17125-17130.
- [7] T. Yang, X. Zhao, Q. Xu, C. Zhou, L. He, S. Nagase. Non-IPR endohedral fullerene Yb@C₇₆: density functional theory characterization. *J. Mater. Chem.* **2011**. Vol.21. P.12206-12209.
- [8] Y. Hao, L. Feng, W. Xu, Z. Gu, Z. Hu, Z. Shi, Z. Slanina, F. Uhlík. Sm@C_{2v}(19138)-C₇₆: a non-IPR cage stabilized by a divalent metal ion. *Inorg. Chem.* **2015**. Vol.54(9). P.4243-4248.
- [9] F. Liu, S. Wang, C.-L. Gao, Q. Deng, X. Zhu, A. Kostanyan, R. Westerström, F. Jin, S.-Y. Xie, A.A. Popov, T. Greber, S. Yang. Mononuclear clusterfullerene single-molecule magnet containing strained

- fused-pentagons stabilized by a nearly linear metal cyanide cluster. *Angew. Chem., Int. Ed.* **2017**. Vol.56. P.1830-1834.
- [10] W. Cai, L. Abella, J. Zhuang, X. Zhang, L. Feng, Y. Wang, R. Morales-Martínez, R. Esper, M. Boero, A. Metta-Magaña, A. Rodríguez-Fortea, J.M. Poblet, L. Echegoyen, N. Chen. Synthesis and characterization of non-Isolated-Pentagon-Rule actinide endohedral metallofullerenes $U@C_{1(17418)}-C_{76}$, $U@C_{1(28324)}-C_{80}$, and $Th@C_{1(28324)}-C_{80}$: low-symmetry cage selection directed by a tetravalent ion. *J. Am. Chem. Soc.* **2018**. Vol.140. P.18039-18050.
- [11] I.N. Ioffe, A.A. Goryunkov, N.B. Tamm, L.N. Sidorov, E. Kemnitz, S.I. Troyanov. Fusing pentagons in a fullerene cage by chlorination: IPR D_2-C_{76} rearranges into non-IPR $C_{76}Cl_{24}$. *Angew. Chem., Int. Ed.* **2009**. Vol.48. P.5904-5907.
- [12] S.M. Sudarkova, O.N. Mazaleva, R.A. Konoplev-Esgenburg, S.I. Troyanov, I.N. Ioffe. Versatility of chlorination-promoted skeletal transformation pathways in C_{76} fullerene. *Dalton Trans.* **2018**. Vol.47. P.4554-4559.
- [13] S. Yang, I.N. Ioffe, S.I. Troyanov. Chlorination-promoted skeletal transformations of fullerenes. *Acc. Chem. Res.* **2019**. Vol.52. P.1783-1792.
- [14] V.I. Kovalenko, A.R. Khamatgalimov. Regularities in the molecular structures of stable fullerenes. *Russ. Chem. Rev.* **2006**. Vol.75. No.11. P.981- 988.
- [15] A.R. Khamatgalimov, V.I. Kovalenko. Molecular structures of unstable isolated-pentagon-rule fullerenes $C_{72}-C_{86}$. *Russ. Chem. Rev.* **2016**. Vol.85. No.8. P.836-853.
- [16] V.I. Kovalenko, A.R. Khamatgalimov. Structure and stability of higher fullerenes. *Russian Academy of Sci. Publ., Moscow.* **2019**. 212 p. (russian)
- [17] R.A. Tuktamysheva, A.R. Khamatgalimov, and V.I. Kovalenko. Electronic and geometric structure of a number of fullerene isomers C_{90} and structure of their chlorine and perfluoroalkyl polyadducts. *Butlerov Communications.* **2014**. Vol.37. No.1. P.1-12. ROI-jbc-02/14-37-1-1
- [18] M.V. Mitroshkina, A.R. Khamatgalimov, and V.I. Kovalenko. Molecular structure of the two IPR isomers 234 (C_s) and 258 (C_1) of a higher fullerene C_{104} . *Butlerov Communications.* **2017**. Vol.49. No.1. P.28-33. DOI: 10.37952/ROI-jbc-01/17-49-1-28
- [19] A.A. Gaynullina, A.R. Khamatgalimov, and V.I. Kovalenko. The structure and stability of higher fullerenes C_{104} : IPR isomers 812 (D_2) and 822 (D_{3d}) *Butlerov Communications.* **2017**. Vol.49. No.3. P.75-83. DOI: 10.37952/ROI-jbc-01/17-49-3-75
- [20] Z. Slanina, F. Uhlík, L. Feng, L. Adamowicz. Evaluation of the relative stabilities of two non-IPR isomers of $Sm@C_{76}$. *Fuller. Nanotub. Car. Nanostruct.* **2016**. Vol.24(5). P.339-344.
- [21] F.-W. Gao, H.-L. Xu, Z.-M. Su. The inner-induced effects of YCN in C_{76} on the structures and nonlinear optical properties. *J. Mol. Model.* **2016**. Vol.22. P.174 (1-7).
- [22] A.R. Khamatgalimov, and V.I. Kovalenko. Structure of non-IPR isomer 4169 (C_s) of C_{66} fullerene and the reason of its stabilization as derivatives. *Butlerov Communications.* **2012**. Vol.32. No.12. P.141-148. ROI-jbc-01/12-32-12-141
- [23] A.R. Khamatgalimov, V.I. Kovalenko. The structure of fullerene C_{66} , which does not obey the Rule of Isolated Pentagons, and endohedral metallofullerene $Sc_2@C_{66}$: quantum-chemical calculations. *Russ. J. Phys. Chem. A.* **2008**. Vol.82. P.1164-1169. (russian)
- [24] A.R. Khamatgalimov, S.S. Korolev, A.A. Arkhipov, A.I. Arkhipov, V.I. Kovalenko. Stability of the non-IPR isomers 6140 (D_3) and 6275 (D_3) of fullerene C_{68} . *Fuller. Nanotub. Car. Nanostruct.* **2008**. Vol.16. P.542-545.
- [25] A.D. Becke. Density-functional thermochemistry. III. The role of exact exchange. *J. Chem. Phys.* **1993**. Vol.98. P.5648-5652.
- [26] C. Lee, W. Yang, R.G. Parr. Development of the Colle-Salvetti correlation-energy formula into a functional of the electron density. *Phys. Rev. B.* **1988**. Vol.37. P.785-789.
- [27] Gaussian 09 / M.J. Frisch, G.W. Trucks [et. al]; *Gaussian, Inc. Wallingford CT.* **2009**.
- [28] V.I. Kovalenko, A.R. Khamatgalimov. Open-shell fullerene C_{74} : phenalenyl-radical substructures. *Chem. Phys. Lett.* **2003**. Vol.377. No.3-4. P.263-268.
- [29] A.R. Khamatgalimov, M. Melle-Franco, A.A. Gaynullina, V.I. Kovalenko. Ythrene: From the Real Radical Fullerene Substructure to Hypothetical (yet?) Radical Molecules. *J. Phys. Chem. C.* **2019**. Vol.123. No.3. P.1954-1959.
- [30] A.R. Khamatgalimov, V.I. Kovalenko. Electronic structure and stability of fullerene C_{82} IPR isomers. *J. Phys. Chem. A.* **2011**. Vol.115(44). P.12315-12320.
- [31] K. Hedberg, L. Hedberg, D.S. Bethune, C.A. Brown, H.C. Dorn, R.D. Johnson, de Vries M. Bond lengths in free molecules of Buckminsterfullerene, C_{60} , from gas-phase electron diffraction. *Science.* **1991**. Vol.254. P.410-412.

- [32] K. Hedberg, L. Hedberg, M. Buhl, D.S. Bethune, C.A. Brown, R.D. Johnson. Molecular structure of free molecules of the fullerene C₇₀ from gas-phase electron diffraction. *J. Am. Chem. Soc.* **1997**. Vol.119. P.5314.
- [33] V.I. Kovalenko, R.A. Tuktamysheva, A.R. Khamatgalimov. Electronic structures of some of C₈₄ fullerene isomers and the structures of their perfluoroalkyl derivatives. *Russ. J. Phys. Chem. A.* **2014**. Vol.88 (1). P.103-107. (russian)
- [34] R.A. Tuktamysheva. Molecular structure of a row of isomers of fullerenes C₈₄ and C₉₀: the relationship with regioisomerism of their perfluoroalkyl and chlorine derivatives: *Dis. ... Cand. Chem. Sciences: 02.00.04*. Kazan. **2015**. 189p. (russian)
- [35] S. Yang, C. Chen, T. Wie, N.B. Tamm, E. Kemnitz, S.I. Troyanov. X-ray crystallographic proof of the isomer D₂-C₈₄(5) as trifluoromethylated and chlorinated derivatives, C₈₄(CF₃)₁₆, C₈₄Cl₂₀, and C₈₄Cl₃₂. *Chem. Eur. J.* **2012**. Vol.18. P.2217-2220.
- [36] I.E. Kareev, I.V. Kuvychko, N.B. Shustova, S.F. Lebedkin, V.P. Bubnov, O.P. Anderson, A.A. Popov, O.V. Boltalina, S.H. Strauss. C₁-(C₈₄-C₂(11))(CF₃)₁₂: trifluoromethylation yields structural proof of a minor C₈₄ cage and reveals a principle of higher fullerene reactivity. *Angew. Chem. Int. Ed.* **2008**. Vol.47. P.6204-6207.